

# Facile Synthesis and Characterization of series of different 2,5 disubstituted 5-(quinolin-2-yl)-oxazoles

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#### ABSTRACT

An efficient method to 2,5-disubstituted oxazoles**3a-i** is developed *via* a 1,3-dipolar cycloaddition/ring-cleavage/1,2-Hmigration/denitrogenation/Cu-catalysed aerobic oxidative dehydrogenative cyclization. The desired products can beattained from readily available 2-vinylquinoline **1** and diverseazides**2a-i**using air as the oxidant beneath conditions, and it offers an attractive substitute method for the synthesis of oxazole derivatives which are further confirmed by spectral analysis.



Scheme I Synthesis of 5-(quinolin-2-yl)-2-substituted oxazoles 3a-i

**KEYWORDS**Oxazole,1,3-dipolar cycloaddition, 2-vinylquinoline, azide.

## **INTRODUCTION**

Nitrogen heterocycles embody the core structure of many drugentrants with a broad spectrum of pharmaceutical and therapeutic outlooks [1–6]. The biological budding of quinoline spinoffs has been previously [7-8].

Peptides play crucial and diverse biological roles as signalling and regulatory molecules in numerousphysiological and pathological processes, such as immunity, stress, growth, homeostasis, reproduction, and other cell functions [9]. Synthesis, bio-specificity, and efficacyprofile in people, peptides provide a stand for the design of novel therapeutic drugs [10-12].

A systematic search for peptidomimetics asoxazole-based peptides [13-14]. Oxazoles are an important class of 5-membered N,O-heterocyclic compounds with numerous applications, from medicines to agrochemical products. The presence of oxazolemoieties in natural peptides confers stability and electronic dispersal to the peptide chain, thereby qualifying peptide–protein gratitude and DNA/RNA–peptide interactions [15].



Furthermore, oxazolesshow anti-bacterial [16-17], anti-viral [18], anti-malarial [19], and anti-algal properties [20]. Herein, we design, synthesis and characterization of target is shown below.



Designed Target 3a-i

## **EXPERIMENTAL SECTION**

#### Material and Methods

All the commercially available chemicals and reagents were further used without purification. The purity of the compounds was analyzed by TLC using Merck 60F254 silica gel plates. The <sup>1</sup>H &<sup>13</sup>CNMR spectra recorded with a Mercury Plus spectrometer had chemical shifts that were referenced to TMS. ESI mass spectra were obtained using a Shimadzu QP5050A quadrupole-based mass spectrometer. Elemental analyses were performed on Carlo Ebra 106 and Perkin Elmer Model 240 analyzers.Synthesis of 5-(quinolin-2-yl)-2-substituted oxazoles designed targets 3a-iare depicted in Schem I.

## **RESULTS AND DISCUSSIONS**

We initiated our research on the model reaction of 2-vinylquinoline (1)with benzyl azide (2a) under different reaction conditions (Table 1).Under an atmosphere of air, the 2,5-disubstituted oxazole (3a) treatment of a 1: 1.2 mixture of 1 and 2a with 10 mol % CuCl in toluene at 80°C for 8 h. It is worth noting that 1 equiv. of the CuCl was sufficient to promote the reaction[21]effectively, and the yield of 3a

Table 1 Optimization of reaction conditions<sup>a</sup>





Entry	Catalyst	Solvent	Yield (%) <sup>b</sup> of 3a		
1	CuCl (10 mol %)	toluene	70		
2	CuBr (10 mol %)	toluene	68		
3	CuI (10 mol %)	toluene	64		
4	none	toluene	0		
5	CuCl (1 equiv.)	toluene	80		
Reaction conditions: 2-vinylquinoline1 (0.5 mmol, 1 equiv.), benzyl azide2a (0.6					

mmol, 1.2 equiv), catalyst (0.05 mmol, 10 mol %), solvent (3.0mL), 80 °C, 8 h under air. <sup>*b*</sup>Isolatedyield of pure product based on **1**. <sup>*c*</sup>CuCl (0.5 mmol, 1 equiv.) wasused.





Entry	Name	MF	ESI
3a	2-phenyl-5-(quinolin-2-yl)oxazole	$C_{18}H_{12}N_2O$	272
3b	5-(quinolin-2-yl)-2-(p-tolyl)oxazole	$C_{19}H_{14}N_2O$	286
3c	2-(4-methoxyphenyl)-5-(quinolin-2-yl)oxazole	$C_{19}H_{14}N_2O_2$	302.
3d	2-(4-chlorophenyl)-5-(quinolin-2-yl)oxazole	$C_{18}H_{11}ClN_2O$	306
3e	2-(4-bromophenyl)-5-(quinolin-2-yl)oxazole	$C_{18}H_{11}BrN_2O$	350
3f	4-(5-(quinolin-2-yl)oxazol-2-yl)aniline	$C_{18}H_{13}N_{3}O$	287
3g	2-(4-isopropylphenyl)-5-(quinolin-2-yl)oxazole	$C_{21}H_{18}N_2O$	314
3h	4-(5-(quinolin-2-yl)oxazol-2-yl)benzonitrile	C <sub>19</sub> H <sub>11</sub> N <sub>3</sub> O	297
3i	2-(4-nitrophenyl)-5-(quinolin-2-yl)oxazole	C <sub>18</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub>	317

# Table 2: Names, MFand ESIof Compounds 3a-3i

# Table 3: <sup>1</sup>H & <sup>13</sup>CNMR data of titled compounds 3a-3i

Entry	<sup>1</sup> H NMR (400 MHz, DMSO) (δppm)	<sup>13</sup> CNMR (100 MHz, DMSO) (δ, ppm)
3a	8.27 (s, 1H), 8.06 (s, 1H), 7.68 (dd, <i>J</i> =	160.89,156.40,149.76,148.77,137.95,134.56
	27.2, 7.5 Hz, 4H), 7.59 (s, 1H), 7.44 (d,	(2C),131.28,130.12,129.77,129.12(2C),127.
	J = 6.4 Hz, 3H), 7.37 (d, $J = 9.0$ Hz,	81(2C),127.17(2C),125.47,113.94.
	1H), 7.25 (s, 1H).	
3b	8.24 (s, 1H), 8.05 (s, 1H), 7.74 (s, 2H),	160.89,156.40,149.76,148.77,141.07,137.95,
	7.64 - 7.41 (m, 4H), 7.36 - 7.17 (m,	134.56(2C),130.10,129.77,129.37(2C),129.0
	3H), 2.37 (s, 3H).	4,127.16(2C), 125.47(2C), 113.94, 21.12.
3c	8.25 (s, 1H), 8.06 (s, 1H), 7.75 - 7.57	162.01,160.89,156.40,149.76,148.77,137.95,
	(m, 5H), 7.46 (s, 1H), 7.23 (s, 1H), 7.05	134.56(2C),130.10, 129.77(2C), 127.24(2C),
	(s, 2H), 3.82 (s, 3H)	125.47(2C), 124.97, 114.52, 113.94, 56.03.
3d	8.25 (s, 1H), 8.06 (s, 1H), 7.71 (d, $J =$	160.89,156.40,149.76,148.77,137.95,136.29,
	12.4 Hz, 2H), 7.58 (s, 1H), 7.53 (s, 2H),	134.56(2C),132.42,130.10,129.77(2C),129.1
	7.44 (s, 3H), 7.20 (s, 1H).	5, 128.32, 127.17(2C), 125.47(2C).
3e	8.25 (s, 1H), 8.06 (s, 1H), 7.71 (d, $J =$	160.89,156.40,149.76,148.77,137.95,134.56,
	12.4 Hz, 2H), 7.58 (s, 1H), 7.53 (s, 2H),	132.65,130.42,130.10,129.77,129.57(2C),12
	7.45 (s, 3H), 7.20 (s, 1H).	7.17(2C), 125.47(2C), 124.76, 113.94.
3f	8.25 (s, 1H), 8.05 (s, 1H), 7.75 (s, 2H),	160.89,156.40,154.90,149.76,148.77,137.95,
	7.61 (s, 1H), 7.43 (s, 1H), 7.39 (s, 2H),	134.56,130.10,129.77(2C),127.83,127.17(2
	7.26 (s, 1H), 6.71 (s, 2H), 3.92 (s, 2H).	C),125.47(2C), 119.34, 116.07, 113.94.
3g	8.27 (s, 1H), 8.04 (s, 1H), 7.75 (s, 2H),	160.89,156.40,149.76,148.77,146.88,137.95,
	7.62 (d, $J = 30.1$ Hz, 3H), 7.43 (s, 3H),	134.56(2C),130.13,129.77(2C),127.60(2C),1
	7.26 (s, 1H), 3.10 (s, 1H), 1.33 (s, 6H).	27.17(2C),125.55(2C),113.94,34.19,23.37
		(2C).
3h	8.25 (s, 1H), 8.06 (s, 1H), 7.84 (d, <i>J</i> =	160.89,156.40,149.76,148.77,137.95,134.56
	7.5 Hz, 2H), 7.78 – 7.70 (m, 4H), 7.58	(2C)132.45,131.46(2C), 130.10, 129.77(2C),
	(s, 1H), 7.43 (s, 1H), 7.25 (s, 1H).	127.16(2C), 125.47,119.12,113.94, 111.40.
3i	8.29 (d, <i>J</i> = 21.0 Hz, 3H), 8.08 (s, 1H),	160.89,156.40,149.76,148.77,137.95,136.29,
	7.92 (s, 2H), 7.83 – 7.67 (m, 2H), 7.60	134.56(2C),132.42,130.10,129.77(2C),129.1
	(d, <i>J</i> = 8.9 Hz, 1H), 7.45 (d, <i>J</i> = 15.0	5, 128.32, 127.17(2C), 125.47(2C).
	Hz, 1H), 7.24 (s, 1H).	



## CONCLUSION

Herein, we have described the synthesis of 2,5-disubstituted oxazoles 3a-iwith promising yields, that functions*via* azide-2-vinylquinoline 1,3-dipolar cycloaddition/ring cleavage/1,2-*H* migration/denitrogenation, followed by Cu-catalysed aerobic oxidative dehydrogenative cyclization of the resulting imines. The use of naturally abundant air as an oxidant as well as an oxygen source.

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